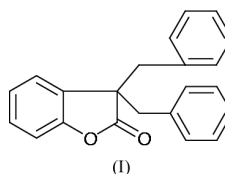


Hai-Bo Wang,\* Jia-Hui Chen and  
Jin-Tang WangDepartment of Applied Chemistry, College of  
Science, Nanjing University of Technology,  
Xinmofan Road No. 5, Nanjing 210009,  
People's Republic of ChinaCorrespondence e-mail:  
wanghaibo@njut.edu.cn

## Key indicators

Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.045  
 $wR$  factor = 0.158  
Data-to-parameter ratio = 13.9For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.3,3-Dibenzylbenzo[*b*]furan-2(3*H*)-oneThe title compound,  $\text{C}_{22}\text{H}_{18}\text{O}_2$ , was synthesized by the  
reaction of methyl (2-hydroxyphenyl)acetate and benzyl  
chloride. In the crystal structure, the dihydrobenzofuran-2-  
one ring system lies on a crystallographic mirror plane.Received 3 September 2004  
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## Comment

2,3-Dihydrobenzofuran-2-one derivatives are of great interest  
because of their biological properties. Some derivatives of 2,3-  
dihydrobenzofuran-2-one have been reported as potential  
inhibitors of angiogenesis (Duflos *et al.*, 2003) and as drugs for  
the treatment of hypoxia (Lepagnol & Lavielle, 1988) and  
some also show high efficacy as intrinsic analgesics and anti-  
inflammatories (Baumann *et al.*, 1986, Rao & Rau, 1985,  
Closse *et al.*, 1981). We report here the crystal structure of the  
title compound, (I).The molecular structure of (I) is shown in Fig. 1; selected  
bond lengths and angles are given in Table 1. Atoms C8–C15,  
O2 and attached H atoms lie on a crystallographic mirror  
plane..

## Experimental

Methyl (2-hydroxyphenyl)acetate (20 mmol) was dissolved in  
acetone (20 ml) and potassium carbonate (30 mmol) was added in  
one portion. Benzyl chloride (40 mmol) was then added to this  
mixture at 293 K. The resulting mixture was refluxed for 20 h. The  
mixture was filtered and the filtrate concentrated under reduced  
pressure to afford crude compound (I). Pure (I) was obtained by  
recrystallization from ethyl acetate (m.p. 454–455 K). Crystals of (I)  
suitable for X-ray diffraction were obtained by slow evaporation of  
an ethanol solution. Spectroscopic analysis,  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , p.p.m.):  
7.20–7.21 (*m*, 1H), 7.12–7.15 (*s*, 8H), 6.93–6.95 (*m*, 4H), 6.70–6.72 (*m*,  
1H), 3.38–3.41 (*m*, 2H), 3.26–3.28 (*m*, 2H).

## Crystal data

 $\text{C}_{22}\text{H}_{18}\text{O}_2$   
 $M_r = 314.36$   
Orthorhombic,  $Pnma$   
 $a = 12.650$  (3) Å  
 $b = 15.488$  (3) Å  
 $c = 8.772$  (2) Å  
 $V = 1718.6$  (6) Å<sup>3</sup>  
 $Z = 4$   
 $D_x = 1.215$  Mg m<sup>-3</sup>Mo  $K\alpha$  radiation  
Cell parameters from 25  
reflections  
 $\theta = 10$ – $13^\circ$   
 $\mu = 0.08$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
Block, colourless  
 $0.4 \times 0.3 \times 0.3$  mm

## Data collection

Nonius CAD-4 diffractometer  
 $\omega/2\theta$  scans  
 Absorption correction:  $\psi$  scan  
 (North *et al.*, 1968)  
 $T_{\min} = 0.973$ ,  $T_{\max} = 0.977$   
 1740 measured reflections  
 1740 independent reflections  
 1002 reflections with  $I > 2\sigma(I)$

$\theta_{\max} = 26.0^\circ$   
 $h = 0 \rightarrow 15$   
 $k = 0 \rightarrow 19$   
 $l = 0 \rightarrow 10$   
 25 standard reflections  
 every 200 reflections  
 intensity decay: none

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.046$   
 $wR(F^2) = 0.158$   
 $S = 0.98$   
 1740 reflections  
 125 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.005$   
 $\Delta\rho_{\max} = 0.20 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.069 (7)

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

|           |           |                       |           |
|-----------|-----------|-----------------------|-----------|
| O1—C9     | 1.194 (3) | C7—C8                 | 1.550 (2) |
| O2—C9     | 1.374 (3) | C8—C15                | 1.503 (3) |
| O2—C10    | 1.404 (3) | C8—C9                 | 1.529 (4) |
| C4—C7     | 1.503 (3) | C8—C7 <sup>i</sup>    | 1.550 (2) |
| C9—O2—C10 | 107.8 (2) | C7—C8—C7 <sup>i</sup> | 109.4 (2) |
| C5—C4—C7  | 121.0 (2) | O1—C9—C8              | 129.4 (2) |
| C4—C7—C8  | 115.1 (2) | O2—C9—C8              | 110.5 (2) |
| C15—C8—C9 | 100.7 (2) | C15—C10—O2            | 111.9 (2) |
| C15—C8—C7 | 113.9 (1) | C10—C15—C8            | 109.1 (2) |
| C9—C8—C7  | 109.3 (1) |                       |           |

Symmetry code: (i)  $x, \frac{1}{2} - y, z$ .

All H atoms were positioned geometrically, with C—H = 0.93–0.97  $\text{\AA}$ , and refined using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier atom})$ .

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97*

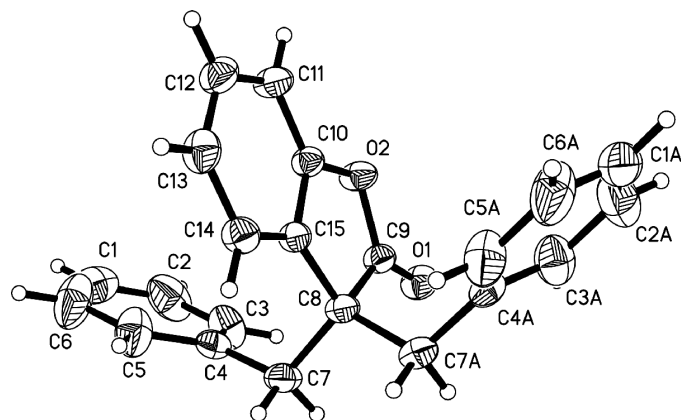


Figure 1

A view of the molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level. [Symmetry code: (A)  $x, \frac{1}{2} - y, z$ .]

(Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

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